



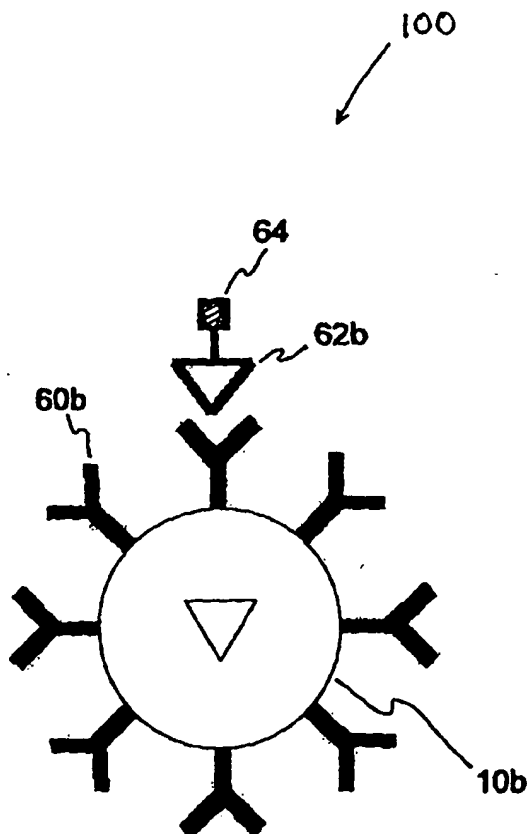
INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁶ : G01N 21/77, 21/64	A1	(11) International Publication Number: WO 98/40726 (43) International Publication Date: 17 September 1998 (17.09.98)
<p>(21) International Application Number: PCT/US98/05025</p> <p>(22) International Filing Date: 13 March 1998 (13.03.98)</p> <p>(30) Priority Data: 08/818,199 14 March 1997 (14.03.97) US</p> <p>(71) Applicant (for all designated States except US): TRUSTEES OF TUFTS COLLEGE [US/US]; Tufts University, Ballou Hall, Medford, MA 02155 (US).</p> <p>(72) Inventors; and (75) Inventors/Applicants (for US only): WALT, David, R. [US/US]; 4 Candlewick Close, Lexington, MA 02178 (US). MICHAEL, Karri, C. [US/US]; 77 Partridge Avenue, Somerville, MA 02145 (US).</p> <p>(74) Agent: CREEHAN, R. Dennis; P.O. Box 750070, Arlington Heights, MA 02175-0070 (US).</p>	<p>(81) Designated States: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GE, GH, GM, GW, HU, ID, IL, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG).</p> <p>Published With international search report. Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</p>	

(54) Title: FIBER OPTIC SENSOR WITH ENCODED MICROSPHERES

(57) Abstract

A microsphere-based analytic chemistry system is disclosed in which microspheres carrying different chemical functionalities may be mixed together while the ability is retained to identify the functionality on each bead using an optically interrogatable encoding scheme. An optical fiber bundle sensor is also disclosed in which the separate microsphere may be optically coupled to discrete fibers or groups of fibers within the bundle. The functionalities are encoded on the separate microspheres using fluorescent dyes and then affixed to wells etched in the end of the bundle. Thus, a single sensor may carry thousands of chemistries. Only those microspheres exhibiting reactions then need to be decoded to identify the corresponding functionality.



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FIBER OPTIC SENSOR WITH ENCODED MICROSPHERES

BACKGROUND OF THE INVENTION

The use of optical fibers and optical fiber strands in combination with light absorbing dyes for chemical analytical determinations has undergone rapid development, particularly within the last decade. The use of optical fibers for such purposes and techniques is described by Milanovich et al., "Novel Optical Fiber Techniques For Medical Application", Proceedings of the SPIE 28th Annual International Technical Symposium On Optics and Electro-Optics, Volume 494, 1980; Seitz, W.R., "Chemical Sensors Based On Immobilized Indicators and Fiber Optics" in C.R.C. *Critical Reviews In Analytical Chemistry*, Vol. 19, 1988, pp. 135-173; Wolfbeis, O.S., "Fiber Optical Fluorosensors In Analytical Chemistry" in *Molecular Luminescence Spectroscopy, Methods and Applications* (S. G. Schulman, editor), Wiley & Sons, New York (1988); Angel, S.M., *Spectroscopy* 2(4):38 (1987); Walt, et al., "Chemical Sensors and Microinstrumentation", ACS Symposium Series, Vol. 403, 1989, p. 252, and Wolfbeis, O.S., *Fiber Optic Chemical Sensors*, Ed. CRC Press, Boca Raton, FL, 1991, 2nd Volume.

When using an optical fiber in an *in vitro/in vivo* sensor, one or more light absorbing dyes are located near its distal end. Typically, light from an appropriate source is used to illuminate the dyes through the fiber's proximal end. The light propagates along the length of the optical fiber; and a portion of this propagated light exits the distal end and is absorbed by the dyes. The light absorbing dye may or may not be immobilized; may or may not be directly attached to the optical fiber itself; may or may not be suspended in a fluid sample containing one or more

analytes of interest; and may or may not be retainable for subsequent use in a second optical determination.

Once the light has been absorbed by the dye, some light of varying wavelength and intensity returns, conveyed through either the same fiber or collection fiber(s) to a detection system where it is observed and measured. The interactions between the light conveyed by the optical fiber and the properties of the light absorbing dye provide an optical basis for both qualitative and quantitative determinations.

Of the many different classes of light absorbing dyes which conventionally are employed with bundles of fiber strands and optical fibers for different analytical purposes are those more common compositions that emit light after absorption termed "fluorophores" and those which absorb light and internally convert the absorbed light to heat, rather than emit it as light, termed "chromophores."

Fluorescence is a physical phenomenon based upon the ability of some molecules to absorb light (photons) at specified wavelengths and then emit light of a longer wavelength and at a lower energy. Substances able to fluoresce share a number of common characteristics: the ability to absorb light energy at one wavelength λ_{ab} ; reach an excited energy state; and subsequently emit light at another light wavelength, λ_{em} . The absorption and fluorescence emission spectra are individual for each fluorophore and are often graphically represented as two separate curves that are slightly overlapping. The same fluorescence emission spectrum is generally observed irrespective of the wavelength of the exciting light and, accordingly, the wavelength and energy of the exciting light may be varied within limits; but the light emitted by the

fluorophore will always provide the same emission spectrum. Finally, the strength of the fluorescence signal may be measured as the quantum yield of light emitted. The fluorescence quantum yield is the ratio of the number of photons emitted in comparison to the number of photons initially absorbed by the fluorophore. For more detailed information regarding each of these characteristics, the following references are recommended: Lakowicz, J. R., *Principles of Fluorescence Spectroscopy*, Plenum Press, New York, 1983; Freifelder, D., *Physical Biochemistry*, second edition, W. H. Freeman and Company, New York, 1982; "Molecular Luminescence Spectroscopy Methods and Applications: Part I" (S.G. Schulman, editor) in *Chemical Analysis*, vol. 77, Wiley & Sons, Inc., 1985; *The Theory of Luminescence*, Stepanov and Gribkovskii, Iliffe Books, Ltd., London, 1968.

In comparison, substances which absorb light and do not fluoresce usually convert the light into heat or kinetic energy. The ability to internally convert the absorbed light identifies the dye as a "chromophore." Dyes which absorb light energy as chromophores do so at individual wavelengths of energy and are characterized by a distinctive molar absorption coefficient at that wavelength. Chemical analysis employing fiber optic strands and absorption spectroscopy using visible and ultraviolet light wavelengths in combination with the absorption coefficient allow for the determination of concentration for specific analytes of interest by spectral measurement. The most common use of absorbance measurement via optical fibers is to determine concentration which is calculated in accordance with Beers' law; accordingly, at a single absorbance wavelength, the greater the quantity of the composition which absorbs light energy at a given wavelength, the greater the optical density

for the sample. In this way, the total quantity of light absorbed directly correlates with the quantity of the composition in the sample.

Many of the recent improvements employing optical fiber sensors in both qualitative and quantitative analytical determinations concern the desirability of depositing and/or immobilizing various light absorbing dyes at the distal end of the optical fiber. In this manner, a variety of different optical fiber chemical sensors and methods have been reported for specific analytical determinations and applications such as pH measurement, oxygen detection, and carbon dioxide analyses. These developments are exemplified by the following publications: Freeman, et al., *Anal Chem.* 53:98 (1983); Lippitsch et al., *Anal. Chem. Acta.* 205:1, (1988); Wolfbeis et al., *Anal. Chem.* 60:2028 (1988); Jordan, et al., *Anal. Chem.* 59:437 (1987); Lubbers et al., *Sens. Actuators* 1983; Munkholm et al., *Talanta* 35:109 (1988); Munkholm et al., *Anal. Chem.* 58:1427 (1986); Seitz, W. R., *Anal. Chem.* 56:16A-34A (1984); Peterson, et al., *Anal. Chem.* 52:864 (1980); Saari, et al., *Anal. Chem.* 54:821 (1982); Saari, et al., *Anal. Chem.* 55:667 (1983); Zhujun et al., *Anal. Chem. Acta.* 160:47 (1984); Schwab, et al., *Anal. Chem.* 56:2199 (1984); Wolfbeis, O.S., "Fiber Optic Chemical Sensors", Ed. CRC Press, Boca Raton, FL, 1991, 2nd Volume; and Pantano, P., Walt, D.R., *Anal. Chem.*, 481A-487A, Vol. 67, (1995).

More recently, fiber optic sensors have been constructed that permit the use of multiple dyes with a single, discrete fiber optic bundle. U.S. Pat. Nos. 5,244,636 and 5,250,264 to Walt, et al. disclose systems for affixing multiple, different dyes on the distal end of the bundle, the teachings of each of these patents being incorporated herein by this reference. The disclosed configurations enable separate optical fibers of the bundle to optically access individual

dyes. This avoids the problem of deconvolving the separate signals in the returning light from each dye, which arises when the signals from two or more dyes are combined, each dye being sensitive to a different analyte, and there is significant overlap in the dyes' emission spectra.

SUMMARY OF THE INVENTION

The innovation of the two previous patents was the placement of multiple chemical functionalities at the end of a single optical fiber bundle sensor. This configuration yielded an analytic chemistry sensor that could be remotely monitored via the typically small bundle. The drawback, however, was the difficulty in applying the various chemistries associated with the chemical functionalities at the sensor's end; the functionalities were built on the sensor's end in a serial fashion. This was a slow process, and in practice, only tens of functionalities could be applied.

The present design is based on two synergistic inventions: 1) the development of a bead-based analytic chemistry system in which beads, also termed microspheres, carrying different chemical functionalities may be mixed together while the ability is retained to identify the functionality of each bead using an optically interrogatable encoding scheme; and 2) an optical fiber bundle sensor in which the separate beads or microspheres may be optically coupled to discrete fibers or groups of fibers within the bundle. Each invention has separate applications but, when implemented together, yields an optical fiber sensor that can support large numbers, thousands or more, of separate chemical functionalities, which is relatively easy to manufacture and use.

In general, according to one aspect, the invention concerns an analytic chemistry system that comprises a

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population of beads or microspheres. Within the population are separate subpopulations, each of which carries chemical functionality which changes an optical signature of the beads in a presence of targeted analytes. This signature change
5 can occur via many different mechanisms. A few examples include the binding of a dye-tagged analyte to the bead, the production of a dye species on or near the beads, the destruction of an existing dye species, a change in optical signal upon analyte interaction with dye on bead, or any
10 other optical interrogatable event. Although the subpopulations may be randomly mixed together, the chemical functionality on each bead is determined via an optical signature which is encoded with a description of the chemical functionality. As a result, by observing whether the optical
15 signature of a particular bead is exhibiting a change, or not, and then decoding the signature for the functionality of the bead, the presence or not of the analyte targeted by the functionality may be determined.

In specific embodiments, the beads are encoded using
20 dyes that are preferably entrapped within the beads, the chemical functionality being added on surfaces. The dyes may be chromophores or phosphors but are preferably fluorescent dyes, which due to their strong signals provide a good signal-to-noise ratio for decoding. The encoding can be
25 accomplished in a ratio of at least two dyes, although more encoding dimensions may be added in the size of the beads, for example.

According to another aspect, the invention also concerns an analytic chemistry fiber optic bundle sensor. This sensor
30 has a population of beads carrying chemical functionalities at, on or near, a distal end of the bundle. The ability to monitor optical signature changes associated with individual or multiple beads is provided by coupling those signature

changes into separate optical fibers or groups of fibers of the bundle for transmission to the proximal end where analysis is performed either manually, by the user, or automatically, using image processing techniques.

5 In the preferred embodiment, each one of the beads is located within separate wells formed at terminal ends of optical fibers of the bundle. These microwells are formed by anisotropic etching of the cores of the optical fibers with respect to the cladding.

10 Also in the preferred embodiment, the population of beads includes separate subpopulations, typically randomly distributed in an array across the bundle end, each subpopulation carrying a different chemical functionality and an optically interrogatable code descriptive of the chemical
15 functionality.

Although each sensor is different insofar that it has a different distribution of the subpopulations of beads within its wells, only those beads that exhibit a positive optical response or signature change need to be decoded. Therefore,
20 the burden is placed on the analysis rather than on sensor manufacture. Moreover, since the beads and fibers in the array can be monodisperse, the fluorescent regions arising from signal generation are extremely uniform and can be analyzed automatically using commercially available
25 microscopy analysis software, such image processing software is capable of defining different spectral regions automatically and counting the number of segments within each region in several seconds.

The above and other features of the invention including
30 various novel details of construction and combinations of parts, and other advantages, will now be more particularly described with reference to the accompanying drawings and pointed out in the claims. It will be understood that the

particular method and device embodying the invention are shown by way of illustration and not as a limitation of the invention. The principles and features of this invention may be employed in various and numerous embodiments without
5 departing from the scope of the invention.

BRIEF DESCRIPTION OF THE DRAWINGS

In the accompanying drawings, reference characters refer to the same parts throughout the different views. The drawings are not necessarily to scale; emphasis has instead
10 been placed upon illustrating the principles of the invention. Of the drawings:

Fig. 1 is a schematic diagram illustrating the optical signature encoding and chemical functionalizing of the microspheres according to the present invention;

15 Fig. 2 is a process diagram describing the preparation, encoding, and functionalizing of the microspheres of the present invention;

Fig. 3 is a schematic diagram illustrating a microsphere system including microspheres with different chemical
20 functionalities and encoded descriptions of the functionalities;

Fig. 4 is a schematic diagram of the inventive fiber optic sensor and associated instrumentation and control system;

25 Fig. 5A and 5B are micrographs illustrating the preferred technique for attaching or affixing the microspheres to the distal end of the optical fiber bundle;

Fig. 6 is a process diagram describing well formation in the optical fiber bundle and affixation of the microspheres
30 in the wells;

Figs. 7A and 7B are micrographs showing the array of microspheres in their corresponding wells prior and subsequent to physical agitation, tapping and air pulsing,

demonstrating the electrostatic binding of the microspheres in the wells;

Figs. 8A, 8B, and 8C are micrographs from alkaline phosphatase microspheres when exposed to fluorescein di-phosphate, at the fluorescein emission wavelength, at an encoding wavelength for DiIC, and at an encoding wavelength for TRC, respectively;

Figs. 9A and 9B are micrographs showing the optical signal from β -galactosidase microspheres when exposed to fluorescein β -galactopyranoside at the fluorescein emission wavelength and at an encoding wavelength for DiIC, respectively; and

Fig. 10A and 10B are micrographs showing the optical response from xrabbit antibody microspheres prior to and post, respectively, exposure to fluorescein labeled antigens.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

1. Microspheres:

Fig. 1 illustrates the construction of a bead or microsphere 10 according to the principles of the present invention. In common with the prior art, the microsphere 10 is given a chemical functionality 12, which is typically applied to the microsphere's surface. The chemical functionality is designed so that in the presence of the analyte(s) to which it is targeted, an optical signature of the microsphere, possibly including region surrounding it, is changed.

The bead or microsphere need not be spherical, irregular beads may be used. They are typically constructed from plastic or ceramic, and bead sizes ranging from nanometers, e.g., 500 nm, to millimeters, e.g., 1 mm, may be used. Moreover, various bead chemistries may be used. Beads or microspheres comprising methylstyrene, polystyrene, acrylic polymer, latex, paramagnetic, thoria sol, carbon, graphite,

and titanium dioxide are various possibilities. "Microsphere Detection Guide" from Bangs Laboratories, Fishers IN. is a helpful guide.

The inventive microsphere has an optical signature that
5 is encoded with a description of the chemical functionality.

In the preferred embodiment, reporter dyes 14 are added to the microsphere 10 with the encoding occurring in the ratio of two or more dyes. The reporter dyes 14 may be chromophore-type. Fluorescent dyes, however, are preferred
10 because the strength of the fluorescent signal provides a better signal-to-noise ratio when decoding. Additionally, encoding in the ratios of the two or more dyes, rather than single dye concentrations, is preferred since it provides insensitivity to the intensity of light used to interrogate
15 the reporter dye's signature and detector sensitivity.

Fig. 2 is a process diagram illustrating the preparation of the microspheres. In step 50, an aliquot of stock microspheres are vacuum filtered to produce a dry cake. In one implementation, microspheres of methylstyrene (87%) and
20 divinylbenzene (13%) are used that have a 3.1 micrometer (μm) diameter.

The microspheres may be purchased with the desired chemical functionalities already present. A large selection of such pre-prepared microspheres are currently available
25 from a number of commercial vendors. Alternatively, "blank" microspheres may be used that have surface chemistries that facilitate the attachment of the desired functionality by the user. Some examples of these surface chemistries for blank microspheres are listed in Table I.

30 The dry cake is then broken apart and a dye solution added to it in step 52 to encode optical signatures of the microspheres with information concerning the intended surface chemical functionalities. Dyes may be covalently bonded to

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the microspheres' surface, but this consumes surface binding sites desirably reserved for the chemical functionalities. Preferably, the microspheres are placed in a dye solution comprising a ratio of two or more fluorescent reporter dyes
5 dissolved in an organic solvent that will swell the microspheres, e.g., dimethylformamide (DMF). The length of time the microspheres are soaked in the dye solution will determine their intensity and the broadness of the ratio
10 range. Longer times yield higher intensities, but broader ratio ranges.

Table I

<u>SURFACE CHEMISTRY</u>	<u>NAME</u>
NH ₂	Amine
COOH	Carboxylic Acid
CHO	Aldehyde
CH ₂ -NH ₂	Aliphatic Amine
CO NH ₂	Amide
CH ₂ -Cl	Chloromethyl
CONH-NH ₂	Hydrazide
OH	Hydroxyl
SO ₄	Sulfate
SO ₃	Sulfonate
Ar NH ₂	Aromatic Amine

In an exemplary two dye system, Texas Red Cadaverine (TRC) is used, which is excited at $\lambda_{ab}=580$ nm and emits at
15 $\lambda_{em}=630$ nm, in combination with indodicarbocyanine (DiIC):

610/670 ($\lambda_{ab}/\lambda_{em}$). Generally, dyes are selected to be compatible with the chemistries involved in the analysis and to be spectrally compatible. The emission wavelengths of the dyes should not overlap the regions of the optical spectrum in which the chemical functionalities induce changes in the microsphere signatures. This avoids deconvolution problems associated with determining signal contributions based on the presence of both the analyte and the encoding dye ratios contributing to an overlapping emission spectral region.

Examples of other dyes that can be used are Oxazin (662/705), IR-144 (745/825), IR-140 (776/882), IR-125 (786/800) from Exiton, and Bodipy 665/676 from Molecular Probes, and Naphthofluorescein (605/675) also from Molecular Probes. Lanthide may also be used. Fluorescent dyes emitting in other than the near infrared may also be used. Chromophore dyes are still another alternative that produce an optically interrogatable signature, as are more exotic formulations using Raman scattering-based dyes or polarizing dyes, for example.

The ability of a particular dye pair to encode for different chemical functionalities depends on the resolution of the ratiometric measurement. Conservatively, any dye pair should provide the ability to discriminate at least twenty different ratios. The number of unique combinations of two dyes made with a particular dye set is shown in the following Table II.

Table II

Number of dyes in set	Combinations possible
3	3
4	6
5	10
6	15

Thus, using six dyes and twenty distinct ratios for each dye pair, 300 separate chemical functionalities may be encoded in a given population of microspheres. Combining more than two dyes provides additional diversity in the encoding combinations.

In step 54, the microspheres are vacuum filtered to remove excess dye. The microspheres are then washed in water or other liquid that does not swell the microspheres, but in which the dyes are still soluble. This allows the residual dye to be rinsed off without rinsing the dye out of the microspheres.

In step 56, the chemical functionality is attached to the microsphere surface chemistries if not already present. It should be understood that surface chemistries may be present throughout the microsphere's volume, and not limited to the physical circumferential surface.

In the prior art, a large spectrum of chemical functionalities have been manifest on microspheres that produce optically interrogatable changes in the presence of the targeted analyte. These functionalities include four broad classifications of microsphere sensors: 1) basic indicator chemistry sensors; 2) enzyme-based sensors; 3) immuno-based sensors; and 3) geno-sensors.

A large number of basic indicator microsphere sensors have been previously demonstrated. Examples include:

Table III

TARGET ANALYTE	CHEMICAL FUNCTIONALITY	NOTES ($\lambda_{ex}/\lambda_{em}$)
pH Sensors based on:	seminaphthofluoresceins	e.g., carboxy-SNAFL
	seminaphthorhodafluors	e.g., carboxy-SNARF
	8-hydroxypyrene-1,3,6-trisulfonic acid	
	fluorescein	

CO ₂ Sensors based On:	seminaphthofluoresceins	e.g., carboxy-SNAFL
	seminaphthorhodafluors	e.g., carbodry-SNARF
	8-hydroxypyrene-1,3,6-trisulfonic acid	
Metal Ions Sensors based on:	desferriozamine B	e.g., Fe
	cyclen derivatives	e.g., Cu, Zn
	derivatized peptides	e.g., FITC-Gly-Gly-His, and FITC-Gly His, Cu, Zn
	fluorexon (calcine)	e.g., Ca, Mg, Cu, Pb, Ba
	calcine blue	e.g., Ca, Mg, Cu
	methyl calcine blue	e.g., Ca, Mg, Cu
	ortho-dianisidine tetracetic acid (ODTA)	e.g., Zn
	bis-salicylidene ethylenediamine (SED)	e.g., Al
	N-(6-methoxy-8-quinolyl)-p-toluenesulfonamine (TSQ)	e.g., Zn
	Indo-1	e.g., Mn, Ni
	Fura-2	e.g., Mn, Ni
	Magnesium Green	e.g., Mg, Cd, Tb
O ₂	Siphenylisobenzofuran	409/476
	Methoxyvinyl pyrene	352/401
Nitrite	diaminonaphthaline	340/377
NO	luminol	355/411
	dihydrohodamine	289/none
Ca ²⁺	Bis-fura	340/380
	Calcium Green	visible light/530
	Fura-2	340/380
	Indo-1	405/485
	Fluo-3	visible light/525
	Rhod-2	visible light/570
Mg ²⁺	Mag-Fura-2	340/380
	Mag-Fura-5	340/380
	Mag-Indo-1	405/485

	Magnesium Green	475/530
	Magnesium Orange	visible light/545
Zn ²⁺	Newport Green	506/535
TSQ	Methoxy-Quinobyl	334/385
Cu ⁺	Phen Green	492/517
Na ⁺	SBFI	339/565
	SBFO	354/575
	Sodium Green	506/535
K ⁺	PBFI	336/557
Cl ⁻	SPQ	344/443
	MQAE	350/460

Each of the chemicals listed in Table III directly produces an optically interrogatable signal or optical signature change in the presence of the targeted analyte.

5 Enzyme-based microsphere sensors have also been demonstrated and could be manifest on microspheres. As shown in Table IV, examples include:

Table IV

SENSOR TARGET	CHEMICAL FUNCTIONALITY
Glucose Sensor	glucose oxidase (enz.) + O ₂ -sensitive dye (see Table I)
Penicillin Sensor	penicillinase (enz.) + pH-sensitive dye (see Table I)
Urea Sensor	urease (enz.) + pH-sensitive dye (see Table I)
Acetylcholine Sensor	acetylcholinesterase (enz.) + pH-sensitive dye (see Table I)

10

Generally, the induced change in the optical signature due to the presence of the enzyme-sensitive chemical analyte

occurs indirectly in this class of chemical functionalities.

The microsphere-bound enzyme, e.g., glucose oxidase, decomposes the target analyte, e.g., glucose, consume a co-substrate, e.g., oxygen, or produce some by-product, e.g.,
5 hydrogen peroxide. An oxygen sensitive dye is then used to trigger the signature change.

Techniques for immobilizing enzymes on microspheres, are known in the prior art. In one case, NH_2 surface chemistry microspheres are used. Surface activation is achieved with a
10 2.5% glutaraldehyde in phosphate buffered saline (10 mM) providing a pH of 6.9. (138 mM NaCl, 2.7 mM, KCl). This is stirred on a stir bed for approximately 2 hours at room temperature. The microspheres are then rinsed with ultrapure water plus 0.01% tween 20 (surfactant) -0.02%, and rinsed
15 again with a pH 7.7 PBS plus 0.01% tween 20. Finally, the enzyme is added to the solution, preferably after being prefiltered using a 0.45 μm amicon micropure filter.

Immuno-based microsphere sensors have been demonstrated for the detection for environmental pollutants such as
20 pesticides, herbicides, PCB's and PAH's. Additionally, these sensors have also been used for diagnostics, such as bacterial (e.g., leprosy, cholera, lyme disease, and tuberculosis), viral (e.g., HIV, herpes simplex, cytomegalovirus), fungal (e.g., aspergillosis, candidiasis,
25 cryptococcoses), Mycoplasmal (e.g., mycoplasmal pneumonia), Protozoal (e.g., amoebiasis, toxoplasmosis), Rickettsial (e.g., Rock Mountain spotted fever), and pregnancy tests.

Microsphere genosensors have also been demonstrated. These are typically constructed by attaching a probe sequence
30 to the microsphere surface chemistry, typically via an NH_2 group. A fluorescent dye molecule, e.g., fluorescein, is attached to the target sequence, which is in solution. The optically interrogatable signature change occurs with the

binding of the target sequences to the microsphere. This produces a higher concentration of dye surrounding the microsphere than in the solution generally. A few demonstrated probe and target sequences, see Ferguson, J.A. et al. *Nature Biotechnology*, Vol. 14, Dec. 1996, are listed below in Table V.

Alternatively, upon binding of the target sequences, an intercalating dye (e.g., ethidium bromide) can be added subsequently to signal the presence of the bound target to the probe sequence.

Table V

PROBE SEQUENCES	TARGET SEQUENCES
B-glo(+) (segment of human B-globin) 5'-NH ₂ -(CH ₂) ₈ -TT TTT TTT TCA ACT TCA TCC ACG TTC ACC-3'	B-glo(+)-CF 5'-Fluorescein-TC AAC GTG GAT GAA GTT C-3'
IFNG(interferon gamma 1) 5'-NH ₂ -(CH ₂) ₈ -T ₁₂ -TGG CTT CTC TTG GCT GTT ACT-3'	IFNG-CF 5'-Fluorescein-AG TAA CAG CCA AGA GAA CCC AAA-3'
IL2(interleukin-2) 5'-NH ₂ -(CH ₂) ₈ -T ₁₂ -TA ACC GAA TCC CAA ACT CAC CAG-3'	IL2-CF 5'-Fluorescein-CT GGT GAG TTT GGG ATT CTT GTA-3'
IL4 (interleukin-4) 5'-NH ₂ -(CH ₂) ₈ -T ₁₂ -CC AAC TGC TTC CCC CTC TGT-3'	IL4-CF 5'-Fluorescein-AC AGA GGG GGA AGC AGT TGG-3'
IL6(interleukin-6) 5'-NH ₂ -(CH ₂) ₈ -T ₁₂ -GT TGG GTC AGG GGT GGT TAT T-3'	IL6-CF 5'-Fluorescein - AA TAA CCA CCC CTG ACC CAA C-3'

Fig. 3 illustrates the construction of an analytic chemistry system 100 from microspheres that have different chemical functionalities. Subpopulations of microspheres are represented as 10a, 10b, 10c carrying respective probe sequences 60a, 60b, 60c, as exemplary functionalities. These subpopulations, once manufactured, are combined and mixed with each other. The resulting combined population contains

microspheres with randomly distributed individual chemical functionalities.

Typically, mixing microspheres with different functionalities results in the loss of information regarding the selectivity for each of the corresponding target sequences. In a solution of microspheres with each of the probe sequences 60a, 60b, and 60c, it is possible to determine that at least one of the target sequences 62a, 62b, and 62c is present when a fluorescent marker dye 64 concentration is observed on the microspheres 10. There is no way, however, to determine which probe sequence 60a, 60b, and 60c is generating the activity since the information concerning which microsphere contained which probe sequence was lost when the subpopulations were mixed.

In contrast, in the present invention, each microsphere in each subpopulation is encoded with a common optical signature. In the illustrated example, the subpopulation represented by microsphere 10a has a two reporter dye ratio of 10:1; the subpopulation of microspheres 10b has a ratio of 1:1 of the same reporter dyes, and subpopulation of microspheres 10c has a ratio of 1:10 of the reporter dyes.

Thus, the randomly mixed subpopulations of microspheres are useful as an analytic chemistry system based on each of the carried chemical functionalities 60a-60c separately. The system 100 is added to an analyte of interest to which some of the chemical functionalities may interact. Any interaction changes the optical response of the corresponding microspheres by, for example, binding a fluorescent dye 64 to the microspheres. By identifying the chemical functionalities of the microsphere in which the optical signature has changed, using the encoded dye combinations, information regarding the chemical identity and concentration of the analyte may be gained based upon the interaction or

noninteraction of each functionality contained in the probe 100.

The microspheres exhibiting activity or changes in their optical signature may be identified by utilizing a somewhat "manual" approach of observing the individual microspheres through a microscope. Decoding can also be performed manually, depending on the particular reporter dyes used. It may be helpful to use optical aids such as light filters to observe the light from the microspheres at emission wavelengths of the reporter dyes. While this approach is possible, in the preferred embodiment, the analytic chemistry microsphere system is used with the inventive optical fiber sensor.

2. Optical Fiber Sensor:

Fig. 4 is a schematic block diagram showing the inventive optical fiber sensor 200 and associated control system 210. The fiber optic sensor 200 comprises a fiber optic bundle 202, that is constructed from separately clad fibers so that light does not mix between fibers. The microsphere system 100 is attached to the bundle's distal end 212, with the proximal end 214 being received by a z-translation stage 216 and x-y micropositioner 218. These two components act in concert to properly position the proximal end 214 of the bundle 202 for a microscope objective lens 220. Light collected by the objective lens 220 is passed to a reflected light fluorescence attachment with three pointer cube slider 222. The attachment 222 allows insertion of light from a 75 Watt Xe lamp 224 through the objective lens 220 to be coupled into the fiber bundle 202. The light from the source 224 is condensed by condensing lens 226, then filtered and/or shuttered by filter and shutter wheel 228, and subsequently passes through a ND filter slide 230.

Light returning from the distal end 212 of the bundle 202 is passed by the attachment 222 to a magnification changer 232 which enables adjustment of the image size of the fiber's proximal or distal end. Light passing through the magnification changer 232 is then shuttered and filtered by a second wheel 234. The light is then imaged on a charge coupled device (CCD) camera 236. A computer 238 executes imaging processing software to process the information from the CCD camera 236 and also possibly control the first and second shutter and filter wheels 228, 234. The instrumentation exclusive of the fiber sensor 200, i.e., to the left of the proximal end of the bundle 202 is discussed more completely by Bronk, et al., Anal. Chem. 1995, Vol. 67, number 17, pp. 2750-2752.

The microsphere system 100 may be attached to the distal end of the optical fiber bundle using a variety of compatible processes. It is important that the microspheres are located close to the end of the bundle. This ensures that the light returning in each optical fiber predominantly comes from only a single microsphere. This feature is necessary to enable the interrogation of the optical signature of individual microspheres to identify reactions involving the microsphere's functionality and also to decode the dye ratios contained in those microspheres. The adhesion or affixing technique, however, must not chemically insulate the microspheres from the analyte.

Figs. 5A and 5B are micrographs of the distal end 212 of the bundle 202 illustrating the preferred technique for attaching the microspheres 10 to the bundle 202. Wells 250 are formed at the center of each optical fiber 252 of the bundle 202. As shown in Fig. 5B, the size of the wells 250 are coordinated with the size of the microspheres 10 so that the microspheres 10 can be placed within the wells 250.

Thus, each optical fiber 252 of the bundle 202 conveys light from the single microsphere 10 contained in its well. Consequently, by imaging the end of the bundle 202 onto the CCD array 236, the optical signatures of the microspheres 10 are individually interrogatable.

Fig. 6 illustrates how the microwells 250 are formed and microspheres 10 placed in the wells. A 1 mm hexagonally-packed imaging fiber contains approximately 20,600 individual optical fibers that have cores approximately $3.7\mu\text{m}$ across (Part No. ET26 from Galileo Fibers). Typically, the cores of each fiber are hexagonally shaped as a result of glass hardness and drawing during fiber fabrication. In some cases, the shape can be circular, however.

In step 270, both the proximal and distal ends 212, 214 of the fiber bundle 202 are successively polished on 12 μm , 9 μm , 3 μm , 1 μm , and 0.3 μm lapping films. Subsequently, the ends can be inspected for scratches on an atomic force microscope. In step 272, etching is performed on the distal end 212 of the bundle 202. A solution of 0.2 grams NH_4F (ammonium fluoride) with 600 μl dH_2O and 100 μl of HF (hydrofluoric acid), 50% stock solution, may be used. The distal end 212 is etched in this solution for a specified time, preferably approximately 80 seconds.

Upon removal from this solution, the bundle end is immediately placed in deionized water to stop any further etching in step 274. The fiber is then rinsed in running tap water. At this stage, sonication is preferably performed for several minutes to remove any salt products from the reaction. The fiber is then allowed to air dry.

The foregoing procedure produces wells by the anisotropic etching of the fiber cores 254 favorably with respect to the cladding 256 for each fiber of the bundle 202.

The wells have approximately the diameter of the cores 254, 3.7 μm . This diameter is selected to be slightly larger than the diameters of the microspheres used, 3.1 μm , in the example. The preferential etching occurs because the pure
5 silica of the cores 254 etches faster in the presence of hydrofluoric acid than the germanium-doped silica claddings 256.

The microspheres are then placed in the wells 250 in step 276 according to a number of different techniques. The
10 placement of the microspheres may be accomplished by dripping a solution containing the desired randomly mixed subpopulations of the microspheres over the distal end 212, sonicating the bundle to settle the microspheres in the wells, and allowing the microsphere solvent to evaporate.
15 Alternatively, the subpopulations could be added serially to the bundle end. Microspheres 10 may then be fixed into the wells 250 by using a dilute solution of sulfonated Nafion that is dripped over the end. Upon solvent evaporation, a thin film of Nafion was formed over the microspheres which
20 holds them in place. This approach is compatible for fixing microspheres for pH indication that carry FITC functionality.

The resulting array of fixed microspheres retains its pH sensitivity due to the permeability of the sulfonated Nafion to hydrogen ions. This approach, however, can not be
25 employed generically as Nafion is impermeable to most water soluble species. A similar approach can be employed with different polymers. For example, solutions of polyethylene glycol, polyacrylamide, or polyhydroxymethyl methacrylate (polyHEMA) can be used in place of Nafion, providing the
30 requisite permeability to aqueous species.

An alternative fixation approach employs microsphere swelling to entrap each microsphere 10 in its corresponding microwell 250. In this approach, the microspheres are first

distributed into the microwells 250 by sonicating the microspheres suspended in a non-swelling solvent in the presence of the microwell array on the distal end 212. After placement into the microwells, the microspheres are

5 subsequently exposed to an aqueous buffer in which they swell, thereby physically entrapping them, analogous to muffins rising in a muffin tin.

One of the most common microsphere formations is tentagel, a styrene-polyethylene glycol co-polymer. These

10 microspheres are unswollen in nonpolar solvents such as hexane and swell approximately 20-40% in volume upon exposure to a more polar or aqueous media. This approach is extremely desirable since it does not significantly compromise the diffusional or permeability properties of the microspheres

15 themselves.

Figs. 7A and 7B show polymer coated microspheres 12 in wells 250 after their initial placement and then after tapping and exposure to air pulses. Figs. 7A and 7B illustrate that there is no appreciable loss of microspheres

20 from the wells due to mechanical agitation even without a specific fixing technique. This effect is probably due to electrostatic forces between the microspheres and the optical fibers. These forces tend to bind the microspheres within the wells. Thus, in most environments, it may be unnecessary

25 to use any chemical or mechanical fixation for the microspheres.

In alternative embodiments, additional encoding parameters can be added, such as microsphere size. If a number of sensors needed exceeds a few hundred, it is

30 possible to use microspheres of different sizes to expand the encoding dimensions of the microspheres. Optical fiber arrays can be fabricated containing pixels with different fiber diameters. With different diameters, the largest wells

can be filled with the largest microspheres and then moving onto progressively smaller microspheres in the smaller wells until all size wells are then filled. In this manner, the same dye ratio could be used to encode microspheres of different sizes thereby expanding the number of different oligonucleotide sequences or chemical functionalities present in the array.

EXPERIMENTAL RESULTS

1. Enzyme-Based Sensor:

10 Subpopulation A

Chemical functionality: Alkaline phosphatase

Target substrate: fluorescein diphosphate (FDP)

Reported dye ratio: 1:1 ratio of DiIC:TRC, where DiIC is 1,1',3,3,3',3'-hexamethyl-indodicarbocyanine iodide and TRC is Texas Red cadaverine

A range of ratios of light intensities are selected that are representative of the optical signature for the dye ratio of the subpopulation based on the quantum yield of the two dyes. The optical signature for this subpopulation is:

$$20 \quad \frac{DiIC \lambda \text{ intensity} - \text{ave. DiIC background}}{TRC \lambda \text{ intensity} - \text{ave. TRC background}} = 0.847 \pm 0.23.$$

Subpopulation B

Chemical functionality: B-Galactosidase;

Target substrate: fluorescein di-B-galactopyranoside (FDG)

25 Reporter dye ratio: 10:1 ratio of DiIC:TRC which translates to an optical signature of:

$$\frac{DiIC \lambda \text{ intensity} - \text{ave. DiIC background}}{TRC \lambda \text{ intensity} - \text{ave. TRC background}} = 4.456 \pm 1.27..$$

Subpopulation C

Chemical functionality: B-glucuronidase

30 Target substrate: fluorescein di-B-D-glucuronide (FDGicu).

25

Reporter dye ratio: 1:10 ratio of DiIC:TRC, which translates to an optical signature of:

$$\frac{DiIC \lambda \text{ intensity} - \text{ave. DiIC background}}{TRC \lambda \text{ intensity} - \text{ave. TRC background}} = 0.2136 \pm 0.03..$$

When the microsphere populations are in the presence of one or more of the substrates, the respective enzymes on the microspheres catalyze the breakdown of the substrates producing fluorescein which is fluorescent, emitting light at 530 nanometers when excited at 490 nm. The production of fluorescein localized to particular beads is then monitored. In this approach, the localization of fluorescein around the microspheres is increased by using a substrate solution of 90% glycerol and 10% substrate. The glycerol prevents the generated fluorescein from diffusing away from the microsphere reaction sites.

During the experiment, images in the encoded wavelengths are first taken. Since both DiIC and TRC are excited at 577 nm. Each microsphere's emissions at 670 nm, indicative of the presence of DiIC and 610 nm indicative of the presence of TRC were recorded using a 595 nm dichroic and an acquisition time of 5 seconds for the CCD 236.

Next, the distal end 212 of the fiber bundle is placed in a buffer and another image taken while illuminating the beams with 490 nm light. Emissions in the 530 nm fluorescein wavelengths were recorded with a 505 nm dichroic. In this case, a CCD acquisition time of one second was used. This process provides a background normalizing image. The buffer was removed and the fiber allowed to dry to avoid substrate solution dilution.

The substrate solution is then introduced and CCD images acquired every 30 seconds to a minute for 30 minutes. While illuminating the microspheres with 490 nm light and collecting emissions in the 530 nm range. Fiber is then

placed back in the buffer solution and another background image captured. Those beads that generate a signal indicative of fluorescein production are decoded. Depending in the ratio of the intensity of light from the two reporter
5 dyes, DiIC:TRC, the chemical functionality of the optically active beads may be decoded according to the following table.

0.617 - 1.08	alkaline phosphatase bead
3.188-5.725	β -galactosidase bead
0.183 - 0.243	β -glucunonidese bead

This process is then repeated for the remaining two
10 substrates.

Figs. 8A-8C are images generated by the CCD 236 when the bead populations are exposed to fluorescein di-phosphate. Fig. 8A illustrates the signals from the alkaline phosphatase microspheres when excited at 490 nm and recording emissions
15 at 530 nm, emissions at this wavelength being indicative of fluorescein production. Fig. 8B shows the image captured by the CCD when the microspheres are excited at 577 nm and emissions at 670 nm are recorded. This wavelength is an encoding wavelength indicative of the concentration of DiIC
20 on the microspheres. Finally, Fig. 8C shows the image when the microspheres are excited with 577 nm light and emissions in the 610 nm range are recorded being indicative of the concentration of TRC in the microspheres.

In a similar vein, Figs. 9A and 9B are images when the
25 microspheres are exposed to fluorescein d- β -galactosidase. Fig. 9A shows emissions at 530 nm indicative of the fluorescein production; and Fig. 9B shows light emitted at the 670 nm range indicative of the presence of DiIC.

These micrographs, Fig. 8A-8C and 9A-9B illustrate fluorescein production around the microspheres may be detected as an optical signature change indicative of reactions involving the chemical functionality of the microspheres. The micrographs also illustrate that the optical signatures may be decoded to determine the chemical functionalities on each microsphere.

2. Genosensor:

Three separate subpopulations of beads were used. In subpopulation A, xrabbit antibodies (Ab) were affixed to the surface of the microspheres; in subpopulation B, xgoat antibodies were affixed to the microspheres; and in subpopulation C, xmouse antibodies were affixed to the microspheres. These three separate subpopulations were identified using a DiIC:TRC encoding ratio similar to that in the previously described experiment.

For the first step of the experiment, images at the encoded wavelengths were captured using 577 nm excitation and looking for emissions at 610 and 670 nm. After this decoding, the fiber was placed in a buffer and an image taken at 530 nm with 490 nm excitation. This provided a background normalizing signal at the fluorescein emission wavelength. Next, the fiber was placed in rabbit IgG antigen (Ag) which is fluorescein labeled. Images were then captured every few minutes at the 530 nm emission wavelength for fluorescein. Figs. 10A and 10B are micrographs showing the image captured by the CCD prior to and subsequent to exposure to a rabbit antigen, which clearly show reaction of the selected microspheres within the population.

Note, if the fluorescein background from the antigen solution is too high to see the antibody-antigen signal, the fiber bundle may be placed in a buffer. This removes the background fluorescence leaving only the Ab-Ag signal.

28

While this invention has been particularly shown and described with references to preferred embodiments thereof, it will be understood by those skilled in the art that various changes in form and details may be made therein
5 without departing from the spirit and scope of the invention as defined by the appended claims.

CLAIMS

What is claimed is:

- 5 1. An analytic chemistry system, comprising a population of beads including separate subpopulations, each subpopulation carrying chemical functionality which changes an optical signature of the beads in the presence of targeted analytes, beads in each
10 subpopulation having an optical signature which is encoded with a description of the chemical functionality carried by that subpopulation.
2. The system described in Claim 1, wherein the beads are
15 encoded using dyes.
3. The system described in Claim 2, wherein the dyes are entrapped within the beads and the chemical functionality is on surfaces of the beads.
20
4. The system described in Claim 1, wherein the beads are encoded using fluorescent dyes.
5. The system described in Claim 1, wherein the beads are
25 encoded by controlling a ratio of at least two dyes.
6. The system described in Claim 1, wherein the chemical functionality changes the optical signature by producing an optically active chemical in the presence of targeted
30 analytes.
7. The system described in Claim 1, wherein the optical signature is changed by the chemical functionalities of

30

the beads by the presence or absence of a fluorescent signal.

8. The system described in Claim 1, wherein the chemical
5 functionalities of the beads support sites for hybridization.

9. The system described in Claim 1, wherein the beads are
affixed to a distal end of an optical fiber bundle.

10

10. The system described in Claim 1, wherein the beads are
located within etched wells at terminal ends of optical
fibers of the bundle.

15 11. A chemical analysis method, comprising
preparing separate subpopulations of beads,
each subpopulation carrying chemical
functionalities that change optical signatures of
the beads in the presence of targeted analytes;
20 encoding optical signature of the beads in
each subpopulation with a description of the
chemical functionalities carried by that
subpopulation;
combining the subpopulations to produce a
25 system;
applying the system;
detecting changes in the optical signatures
indicative of a presence of the targeted analytes;
and
30 decoding optical signature of the beads to
identify the chemical functionalities.

12. The method described in Claim 11, wherein encoding the optical signatures with the chemical functionalities comprises doping the beads with fluorescent dyes.
- 5 13. The method described in Claim 11, wherein encoding the optical signatures with chemical functionalities comprises attaching encoding dyes to the beads.
- 10 14. The method described in Claim 11, wherein encoding the optical signatures with the chemical functionalities comprises controlling a ratio of at least two dyes carried by each bead.
- 15 15. The method described in Claim 11, further comprising:
encoding the beads with the chemical
functionalities by entrapping dyes within or
attaching dyes to the beads; and
applying the chemical functionalities to the
beads.
- 20 16. The method described in Claim 11, further comprising enabling the chemical functionalities to produce an optically active species in the presence of targeted analytes to change the optical signature.
- 25 17. The method described in Claim 11, further comprising changing the optical signature by the presence or absence of a fluorescent signal from the beads.
- 30 18. The method described in Claim 11, further comprising enabling the chemical functionalities to hybridize.
19. An analytic chemistry sensor, comprising:

a bundle of optical fibers;
a population of beads carrying chemical
functionalities at a distal end of the fiber optic
bundle, light from individual bead being coupled
5 into separate or groups of separate fibers of the
bundle for transmission to the proximal end of the
bundle.

20. The sensor described in Claim 19, wherein each one of
10 the beads is located within separate wells formed at
terminal ends of optical fibers of the bundle.

21. The sensor described in Claim 20, wherein the wells are
formed by anisotropic etching of the cores of the
15 optical fibers with respect to the cladding.

22. The sensor described in Claim 19, further comprising a
light source for exciting optically active chemicals
bound to the chemical functionalities.

20 23. The sensor described in Claim 19, wherein the population
of beads includes separate subpopulations, each
subpopulation carrying a different chemical
functionality and an optically interrogatable code
25 descriptive of the chemical functionality.

24. The sensor described in Claim 23, further comprising a
light source for exciting optically active chemicals
bound to the chemical functionalities.

30 25. The sensor described in Claim 23, wherein code of each
subpopulation comprises fluorescent dyes.

26. The sensor described in Claim 23, further comprising a filter and a frame capturing camera for detecting optical signatures indicative of a status of the chemical functionalities and optical signatures indicative of the encoding of the beads.
27. A method for constructing and using an analytic chemistry sensor, comprising:
- forming wells at terminal ends of optical fibers within a bundle;
 - distributing beads carrying chemical functionalities within the wells; and
 - monitoring a status of the chemical functionalities from a proximal end of the bundle.
28. The method described in Claim 27, wherein forming the wells comprises anisotropically etching of cores of the optical fibers with respect to cladding.
29. The method described in Claim 27, further comprising forming a population of beads in the wells from separate subpopulations, each subpopulation carrying a different chemical functionality and an optically interrogatable code descriptive of the chemical functionality.
30. The method described in Claim 29, further comprising randomly distributing the subpopulations within the wells.
31. The method described in Claim 29, further comprising serially adding the subpopulations to the wells.

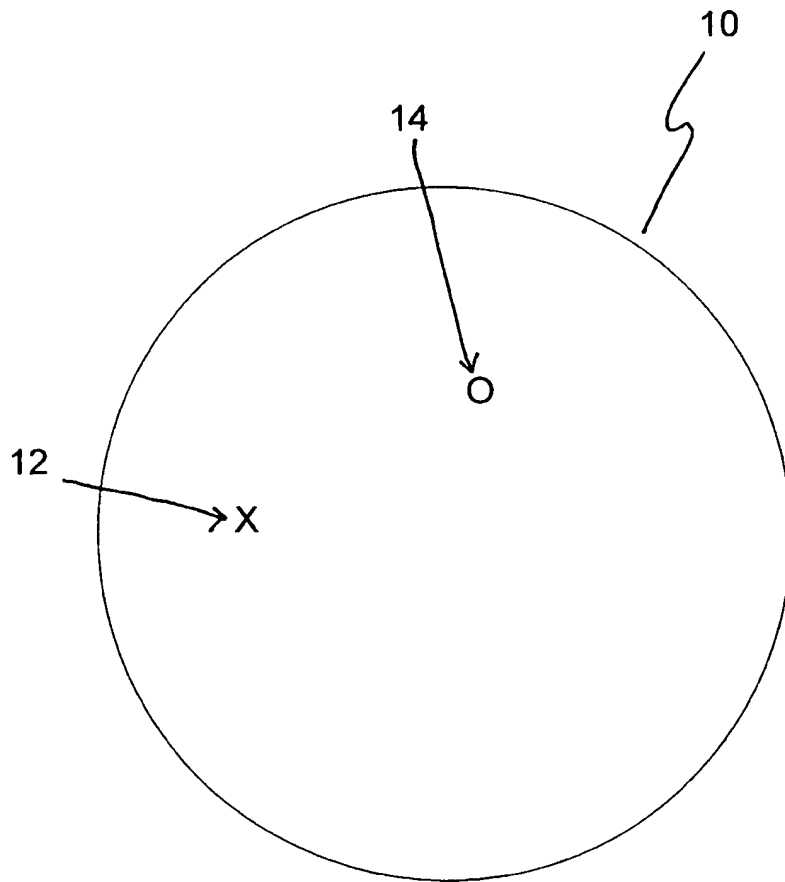


Fig. 1

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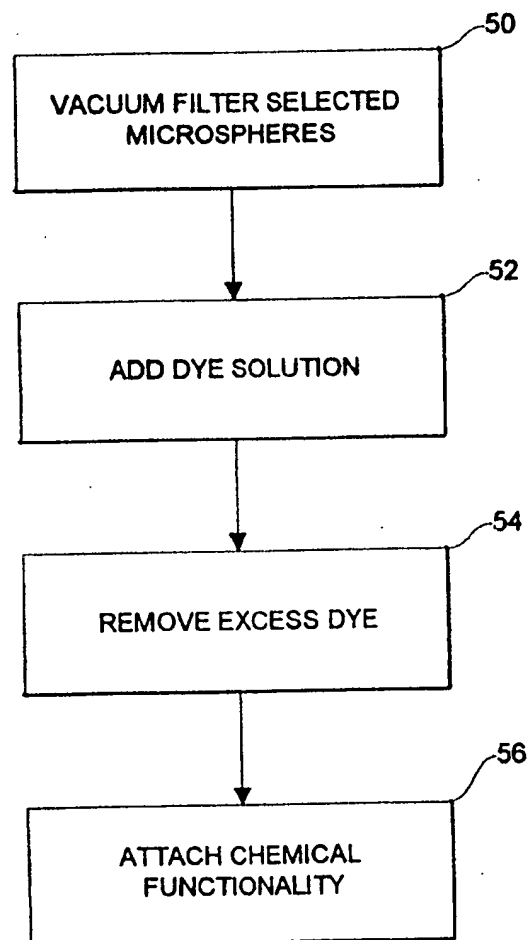


FIG. 2

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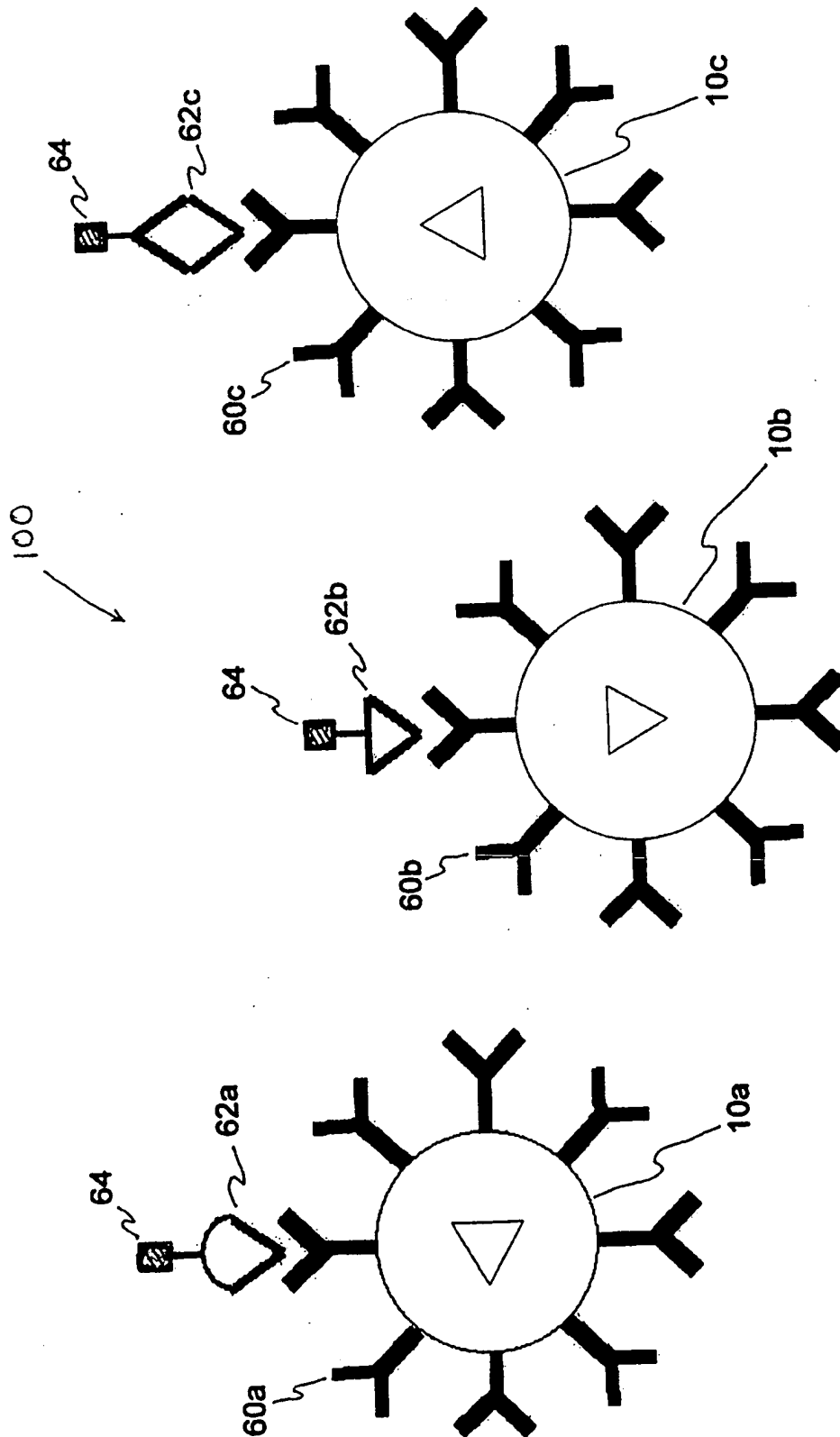


Fig. 3

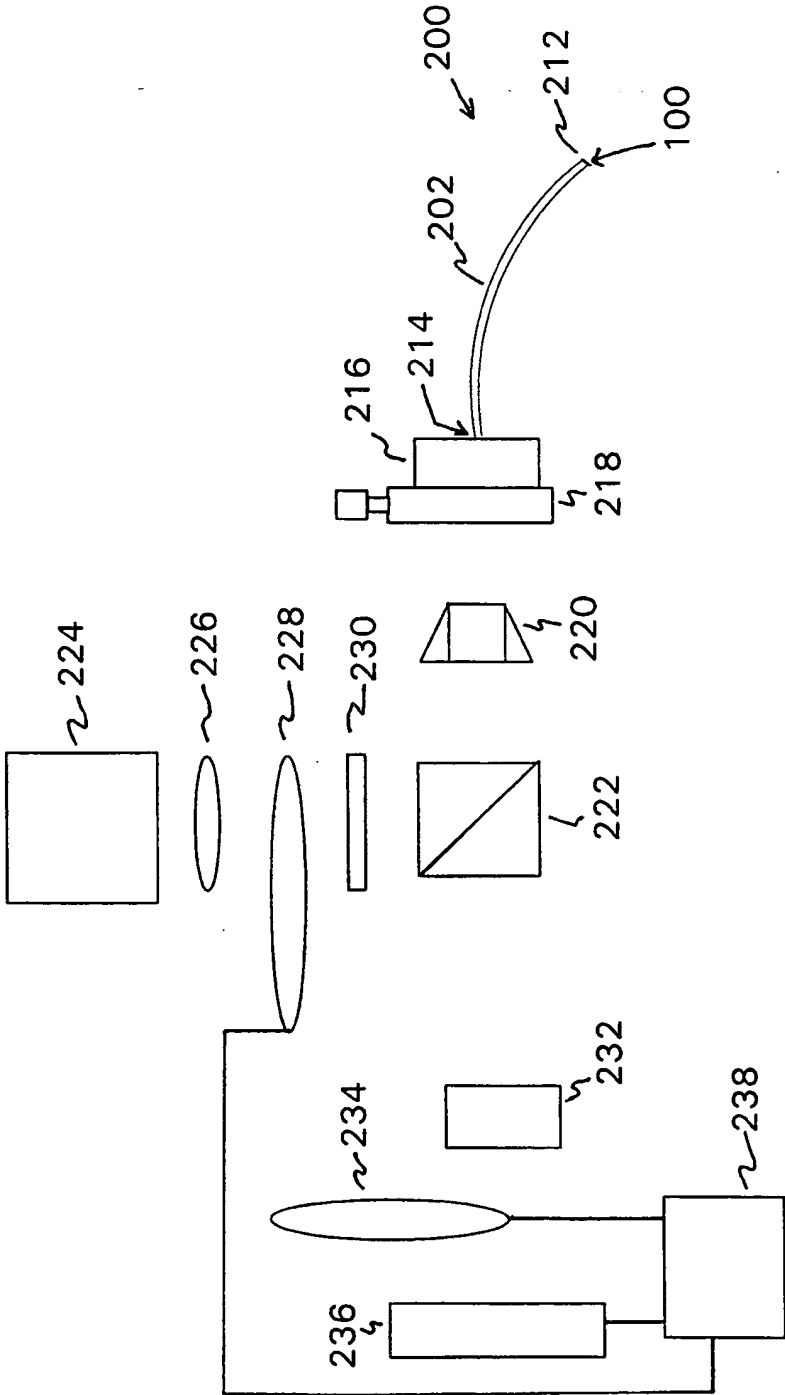
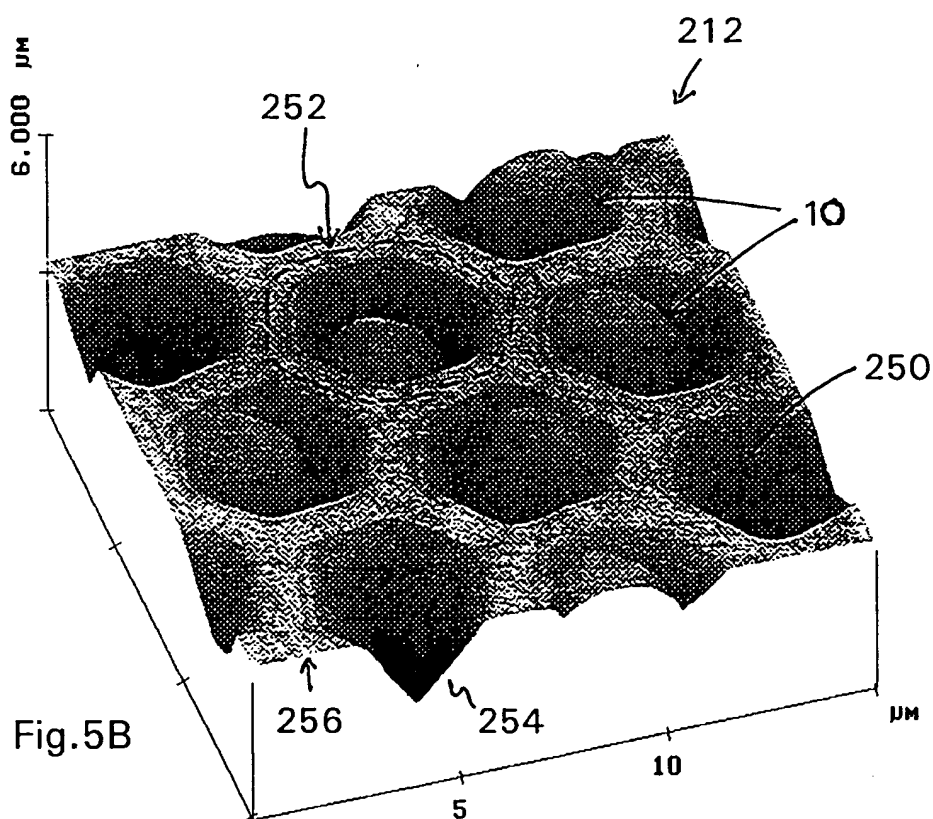
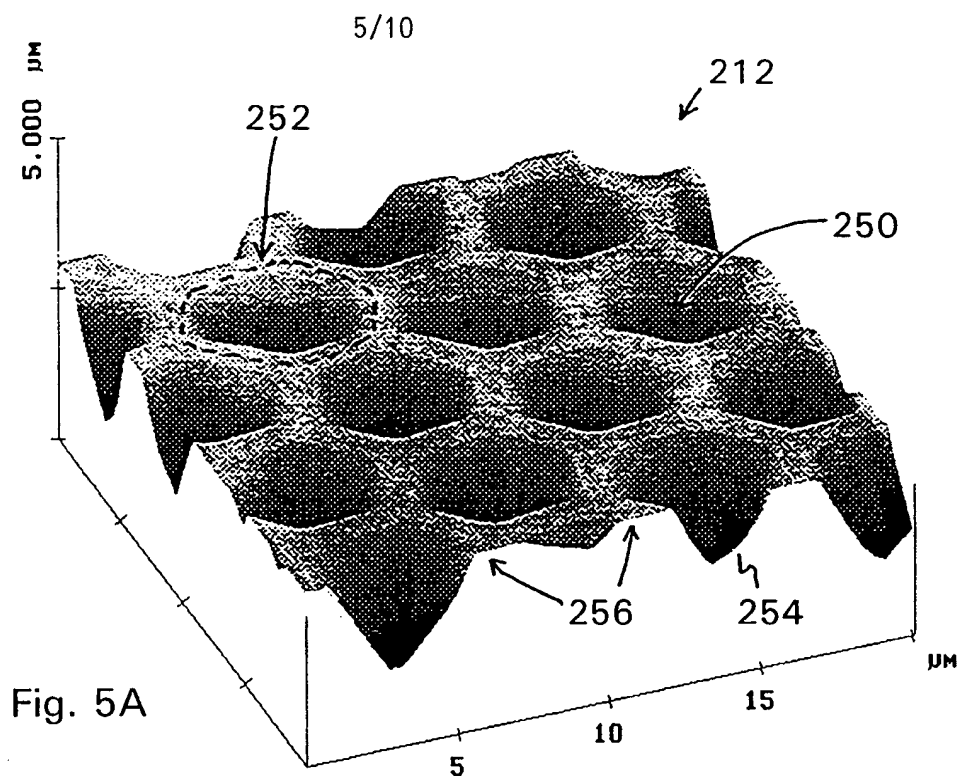


Fig. 4



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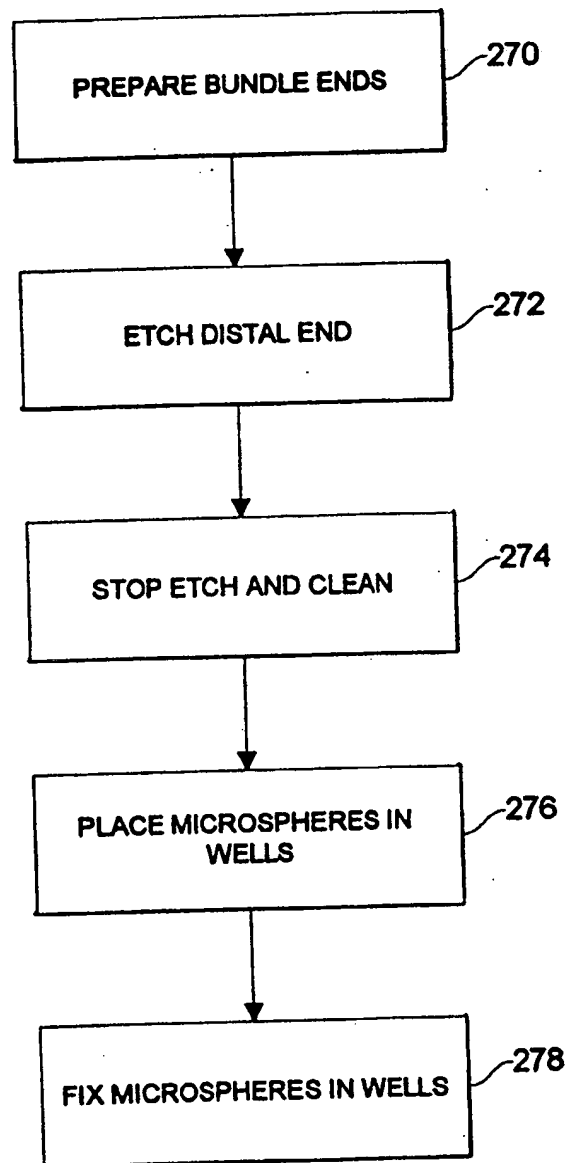


FIG. 6

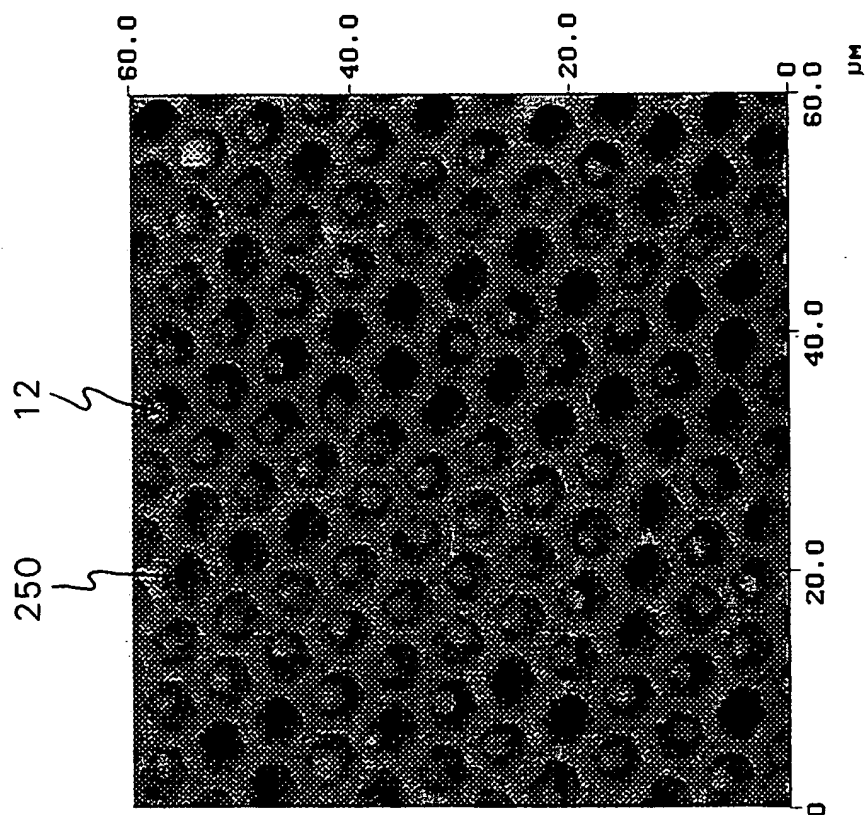


Fig. 7B

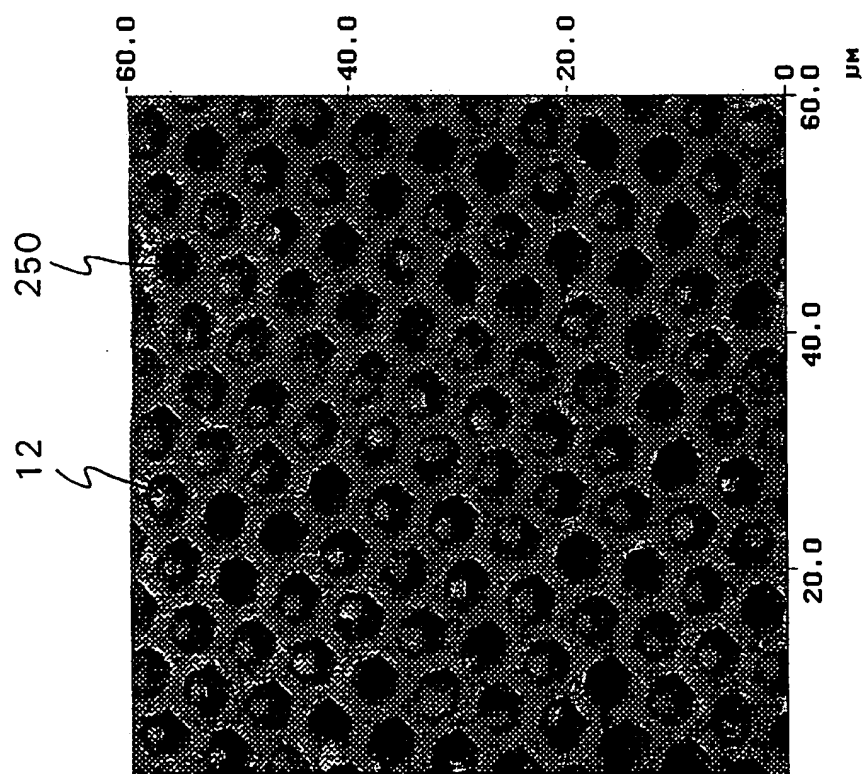


Fig. 7A

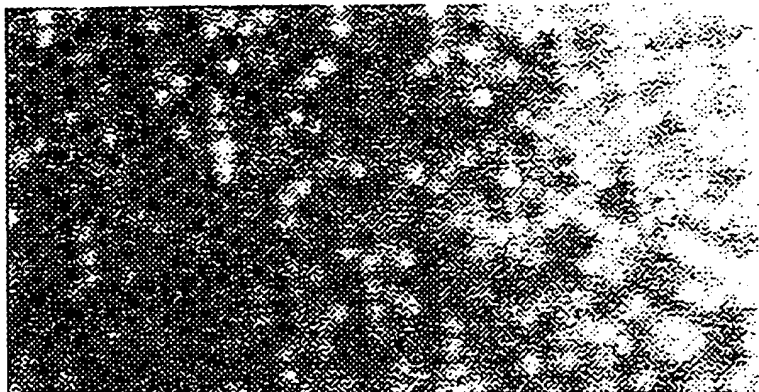


Fig. 8A

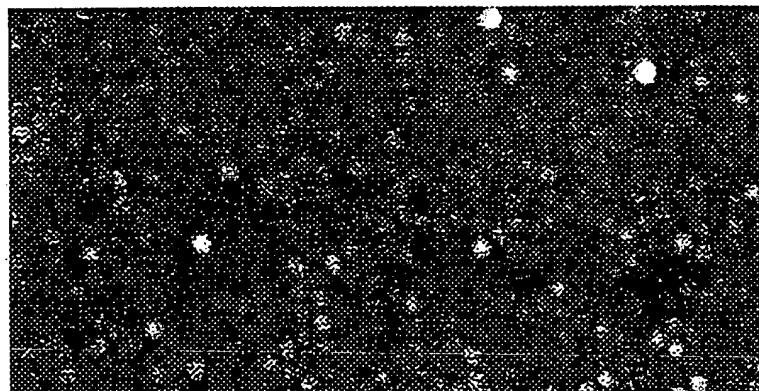


Fig. 8B

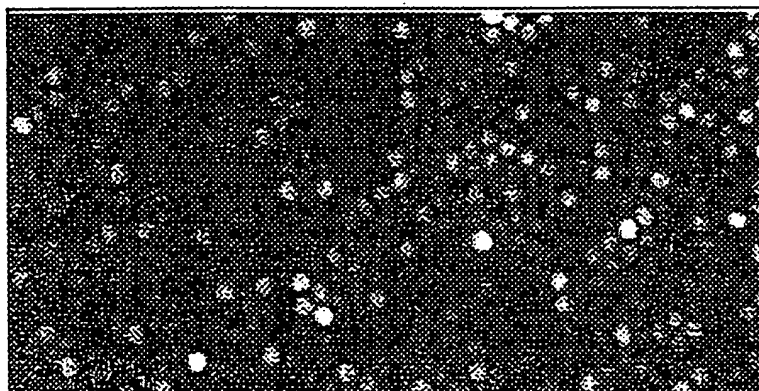


Fig. 8C

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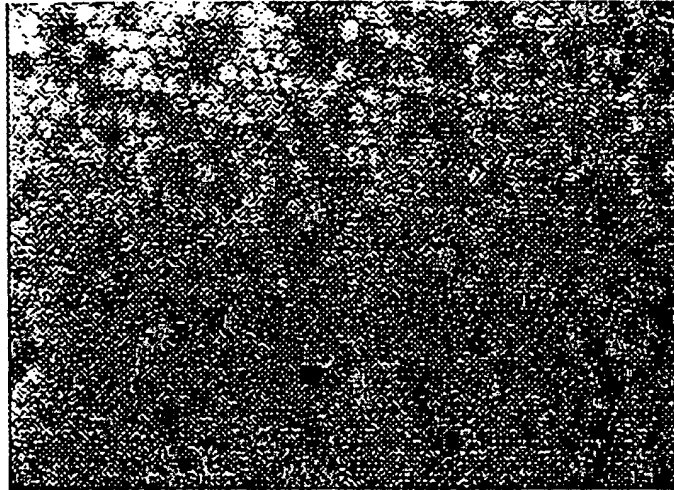


Fig. 9A

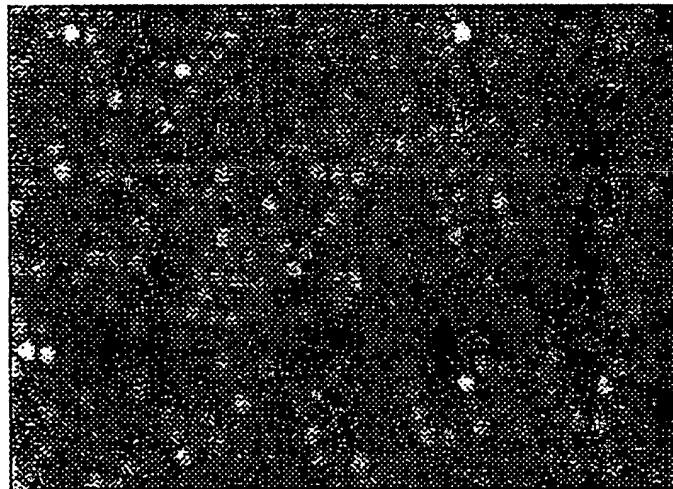


Fig. 9B

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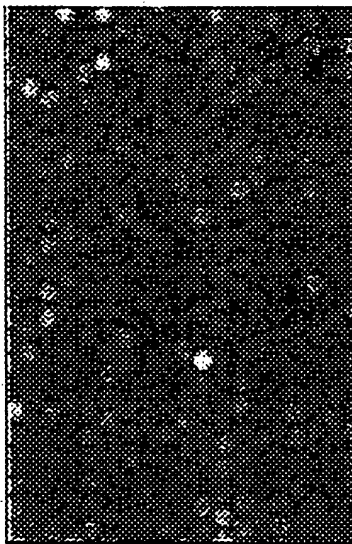


Fig. 10B

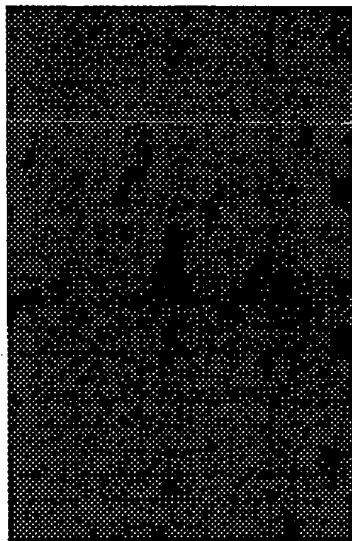


Fig. 10A

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INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 98/05025

A. CLASSIFICATION OF SUBJECT MATTER
IPC 6 G01N21/77 G01N21/64

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 G01N

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X,P	WO 97 14028 A (LUMINEX CORP) 17 April 1997 see abstract	1-8
X	WO 93 02360 A (TRANSMED BIOTECH INCORPORATED) 4 February 1993 see page 4, line 18-31 see page 5, line 6-10 see page 6, line 10-25 see page 6, line 26 - page 7, line 8 see page 7, line 18 - page 8, line 2 see page 9, line 31-32 see page 19, line 36 - page 20, line 2	1-8
X	WO 89 11101 A (MICHAEL JOHN HOLMES) 16 November 1989 see page 5, line 20 - page 6, line 15 see page 7, line 33 - page 8, line 2 see page 8, line 15-20	1-8

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

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Date of the actual completion of the international search

13 August 1998

Date of mailing of the international search report

24/08/1998

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INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 98/05025

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 4 499 052 A (FULWYLER) 12 February 1985 see column 1, line 61 - column 2, line 24 see column 6, line 4-11; examples 1,4 ---	1-8
X	US 5 028 545 A (SOINI) 2 July 1991 see column 1, line 63 - column 2, line 4 see column 2, line 20-37 see column 2, line 62-66 see column 3, line 10-13 see column 3, line 44-49 ---	1-8
A	GB 2 294 319 A (CAMBRIDGE IMAGING LTD) 24 April 1996 see page 8, paragraph 3 - page 9, paragraph 8 see page 14, paragraph 4 see page 42, line 2-6; figures 7A,B,C ---	11,19,27
A	US 4 907 037 A (BOISDE GILBERT ET AL) 6 March 1990 see column 6, line 33 - column 7, line 17; figure 1 ---	11,19,27
A	US 5 308 771 A (ZHOU QUAN ET AL) 3 May 1994 see column 6, line 36-63; figure 2 ---	11,19,27
A	US 5 512 490 A (KAUER JOHN S ET AL) 30 April 1996 see column 34, line 19 - column 36, line 61; figures 22,23 ---	11,19,27
A	EP 0 633 465 A (CONSIGLIO NAZIONALE RICERCHE) 11 January 1995 see column 1, line 1-10 see column 2, line 14 - column 3, line 8 see column 4, line 10-13 ---	11,19,27
A	US 3 791 192 A (BUTLER J) 12 February 1974 see column 9, line 4-26; figure 14 ---	11,19,27
A	US 5 252 294 A (KROY WALTER ET AL) 12 October 1993 see column 2, line 35 - column 3, line 30; figures 1-3 ---	11,19,27
A	EP 0 547 787 A (FLOW VISION INC) 23 June 1993 see column 6, line 13 - column 8, line 38; figures 2,3 -----	11,19,27

INTERNATIONAL SEARCH REPORT

Information on patent family members

national Application No

PCT/US 98/05025

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 9714028 A	17-04-1997	US 5736330 A AU 7398996 A EP 0852004 A	07-04-1998 30-04-1997 08-07-1998
WO 9302360 A	04-02-1993	AU 2348092 A CA 2113350 A EP 0594763 A JP 6509417 T US 5567627 A	23-02-1993 04-02-1993 04-05-1994 20-10-1994 22-10-1996
WO 8911101 A	16-11-1989	AT 127231 T AU 3565889 A CA 1335238 A DE 68924064 D DE 68924064 T EP 0413741 A JP 2742953 B JP 3504276 T US 5585241 A	15-09-1995 29-11-1989 18-04-1995 05-10-1995 08-02-1996 27-02-1991 22-04-1998 19-09-1991 17-12-1996
US 4499052 A	12-02-1985	DE 3331017 A FR 2532431 A GB 2126341 A,B JP 1772176 C JP 4047265 B JP 59060261 A US 4717655 A	08-03-1984 02-03-1984 21-03-1984 14-07-1993 03-08-1992 06-04-1984 05-01-1988
US 5028545 A	02-07-1991	SE 458968 B DE 3870324 A EP 0296136 A JP 1035268 A JP 2638085 B SE 8702511 A	22-05-1989 27-05-1992 21-12-1988 06-02-1989 06-08-1997 17-12-1988
GB 2294319 A	24-04-1996	EP 0787288 A WO 9612942 A	06-08-1997 02-05-1996
US 4907037 A	06-03-1990	FR 2613074 A DE 3889757 D	30-09-1988 07-07-1994

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 98/05025

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 4907037 A		DE 3889757 T EP 0284513 A JP 2597634 B JP 63255645 A	22-12-1994 28-09-1988 09-04-1997 21-10-1988
US 5308771 A	03-05-1994	NONE	
US 5512490 A	30-04-1996	NONE	
EP 0633465 A	11-01-1995	IT 1265878 B US 5530779 A	12-12-1996 25-06-1996
US 3791192 A	12-02-1974	NONE	
US 5252294 A	12-10-1993	DE 3818614 A DE 3825907 A DE 58907327 D EP 0347579 A DE 8817007 U	07-12-1989 01-02-1990 05-05-1994 27-12-1989 02-10-1991
EP 0547787 A	23-06-1993	US 5191388 A BR 9205036 A CA 2083681 A JP 6074891 A	02-03-1993 22-06-1993 19-06-1993 18-03-1994